

09/29/00

JC932 U.S. PTO

10-02-00

A

Please type a plus sign (+) inside this box → ☐

PTO/SB/05 (4/98)

Approved for use through 09/30/2000. OMB 0651-0032

Patent and Trademark Office: U.S. DEPARTMENT OF COMMERCE

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

**UTILITY
PATENT APPLICATION
TRANSMITTAL**

(Only for new nonprovisional applications under 37 C.F.R. § 1.53(b))

Attorney Docket No.	ECB-0004
First Inventor or Application Identifier	Bruce R. Cook, et.al.
Title	CATALYTIC STRIPPING FOR MERCAPTAN REMOVAL
Express Mail Label No.	EL028161475US

APPLICATION ELEMENTS

See MPEP chapter 600 concerning utility patent application contents.

ADDRESS TO: Commissioner for Patents
Box Patent Application
Washington, DC 20231

1. ☒ * Fee Transmittal Form (e.g., PTO/SB/17) (Submit an original and a duplicate for fee processing)
2. ☒ Specification (preferred arrangement set forth below) Total Pages
- Descriptive title of the Invention
- Cross References to Related Applications
- Statement Regarding Fed sponsored R & D
- Reference to Microfiche Appendix
- Background of the Invention
- Brief Summary of the Invention
- Brief Description of the Drawings (if filed)
- Detailed Description
- Claim(s)
- Abstract of the Disclosure
3. ☒ Drawing(s) (35 U.S.C. 113) Total Sheets
4. ☒ Oath or Declaration Total Pages
a. ☒ Newly executed (original or copy)
b. ☐ Copy from a prior application (37 C.F.R. § 1.53(b)) (for continuation/divisional with Box 16 completed)
i. ☐ DELETION OF INVENTOR(S)
Signed statement attached deleting inventor(s) named in the prior application, see 37 C.F.R. §§ 1.63(d)(2) and 1.33(b).
5. ☐ Microfiche Computer Program (Appendix)
6. ☐ Nucleotide and/or Amino Acid Sequence Submission (if applicable, all necessary)
a. ☐ Computer Readable Copy
b. ☐ Paper Copy (identical to computer copy)
c. ☐ Statement verifying identity of above copies

ACCOMPANYING APPLICATION PARTS

7. ☐ Assignment Papers (cover sheet & document(s))
8. ☐ 37 C.F.R. § 3.73(b) Statement (when there is an assignee) ☐ Power of Attorney
9. ☐ English Translation Document (if applicable)
10. ☐ Information Disclosure Statement (IDS)/PTO-1449 ☐ Copies of IDS Citations
11. ☐ Preliminary Amendment (Communication)
12. ☒ Return Receipt Postcard (MPEP 503) (Should be specifically itemized)
13. ☐ * Small Entity Statement(s) (PTO/SB/09-12) ☐ Statement filed in prior application, Status still proper and desired
14. ☐ Certified Copy of Priority Document(s) (if foreign priority is claimed)
15. ☐ Other: _____

* NOTE FOR ITEMS 1 & 13: IN ORDER TO BE ENTITLED TO PAY SMALL ENTITY FEES, A SMALL ENTITY STATEMENT IS REQUIRED (37 C.F.R. § 1.27), EXCEPT IF ONE FILED IN A PRIOR APPLICATION IS RELIED UPON (37 C.F.R. § 1.28).

JC920 U.S. PTO
09/676875

09/29/00

16. FEE CALCULATION

CLAIMS AS FILED				
(1) For	(2) Number Filed	(3) Number Extra	(4) Rate	(5) Basic Fee \$690.00
Total Claims	14 - 20	0	x 18.00	0
Independent Claims	1 - 3	0	x 78.00	0
Multiple Dependent Claim Fee			+ 260.00	0
TOTAL FILING FEE				\$690.00

36926 U.S. PRO
09/676875
09/29/00

- ☐ Fee for recording Assignment (\$40.00).
- ☒ Please charge Deposit Account No. 05-1330 in the amount of \$ 690.00.
- ☒ The Commissioner is hereby authorized to charge any additional fees under 37 CFR 1.16 and 1.17 which may be required during the entire pendency of the application, or credit any overpayment, to Deposit Account No. 05-1330. A duplicate copy of this Form is enclosed.

17. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in a preliminary amendment:

☐ Continuation ☐ Divisional ☐ Continuation-in-part (CIP) of prior application No.: _____ / _____

Prior application information: Examiner: _____ Group / Art Unit: _____

For CONTINUATION or DIVISIONAL APPS only: The entire disclosure of the prior application, from which an oath or declaration is supplied under Box 4b, is considered a part of the disclosure of the accompanying continuation or divisional application and is hereby incorporated by reference. The incorporation can only be relied upon when a portion has been inadvertently omitted from the submitted application parts.

18. CORRESPONDENCE ADDRESS

☐ Customer Number or Bar Code Label (Insert Customer No. or Attach bar code label here) or ☒ Correspondence address below

Name	ESTELLE C. BAKUN				
Address	ExxonMobil Research and Engineering Company (formerly Exxon Research and Engineering Company)				
City	Annandale	State	New Jersey	Zip Code	08801-0900
Country	United States of America	Telephone	(909) 730-3635	Fax	(908) 730-3649

Name (Print / Type)	ESTELLE C. BAKUN	Registration No. (Attorney / Agent)	35,054
Signature	<i>Estelle C. Bakun</i>	Date	9/29/00

☒ "Express Mail" mailing label number **EL028161475US** . Date of Deposit **SEPTEMBER 29, 2000** .

I hereby certify that this paper or fee is being deposited with the United States Postal Service "Express Mail Post Office to Addressee" service under 37 CFR 1.10 on the date indicated above and is addressed to the Commissioner of Patents and Trademarks, Washington, D.C. 20231.

TERESA L. LACHOWSKI

(Typed or printed name of person mailing paper or fee)

Teresa L. Lachowski

(Signature of person mailing paper or fee)

- a -

APPLICATION FOR UNITED STATES PATENT

CATALYTIC STRIPPING FOR MERCAPTAN REMOVAL (ECB-0004)

Applicants: Bruce R. Cook
Robert C. Welch
Richard H. Ernst
William E. Winter, Jr.
Richard A. Demmin
Charles A. Euker, Jr.

"EXPRESS MAIL" mailing label

number

EL028161475US

Date of Deposit September 29, 2000

I hereby certify that this paper or fee is being deposited with the United States Postal Service "Express Mail Post Office to Addressee" service under 37 CFR 1.10 on the date indicated above and is addressed to the Commissioner for Patents, Washington, D.C. 20231.

Teresa L. Lachowski

(Typed or printed name of person mailing paper or fee)

Teresa L. Lachowski

(Signature of person mailing paper or fee)

CATALYTIC STRIPPING FOR MERCAPTAN REMOVAL(ECB-0004)

FIELD OF THE INVENTION

A process is disclosed for decreasing the amount of sulfur in hydrocarbon streams.

BACKGROUND OF THE INVENTION

Environmentally driven regulatory standards for motor gasoline (mogas) sulfur levels will result in the widespread production of 120 ppm S mogas by the year 2004 and 30 ppm by 2006. In many cases, these sulfur levels will be achieved by hydrotreating naphtha produced from Fluid Catalytic Cracking (cat naphtha), which is the largest contributor to sulfur in the mogas pool. As a result, techniques are required that reduce the sulfur in cat naphthas without reducing beneficial properties such as octane.

Conventional fixed bed hydrotreating can reduce the sulfur level of cracked naphthas to very low levels, however, such hydrotreating also results in severe octane loss due to extensive reduction of the olefin content. Selective hydrotreating processes such as SCANfining have recently been developed to avoid massive olefin saturation and octane loss. Unfortunately, in such processes, the liberated H₂S reacts with retained olefins forming mercaptan sulfur by reversion. Such processes can be conducted at severities which produce product within sulfur regulations, however, significant octane loss also occurs.

Several methods exist for removal of sulfur from hydrocarbon streams. For example, U.S. 3,876,532; U.S. 4,149,965; U.S. 5,423,975; and U.S. 5,826,373 each teach hydrotreating methods using deactivated or spent catalyst. U.S. 5,885,440 teaches cooling of a hydrocrackate prior to hydrotreating. U.S.

3,338,819 teaches hydrotreatment of a hydrocrackate over a granular catalyst bed at substantially the same conditions as used to produce the hydrocrackate.

U.S. patents 5,510,016; 5,308,471; 5,399,258; 5,346,609; 5,409,596; and 5,413,697 each teach hydrodesulfurization followed by treatment over an acidic catalyst to restore octane.

What is needed in the art is a process which produces sulfur levels within regulatory amounts and which minimizes loss of product octane.

BRIEF DESCRIPTION OF THE FIGURES

Figure 1 depicts a typical SCANfining process.

Figure 2 depicts one possible embodiment of the invention. The typical SCANfining process flow scheme is included for convenience, however, all that is required in the instant invention is that the stream being treated be previously hydrodesulfurized. Hence, the Figure depicts a previously hydrodesulfurized (SCANfined) feedstream (7) containing mercaptan sulfur entering a three phase reactor (18) with a fixed catalyst bed along with a stripping gas (8) and hydrogen sulfide and gas exiting at (10) and desulfurized product at (9).

Figure 3 depicts one possible embodiment again where the SCANfining step is included merely for convenience. A previously hydrodesulfurized feedstream (7) and stripping gas (21) enter three phase reactor (19). The product from the reactor (19) then undergoes a depressurization step (17) and enters a stripper (18) where gases (10) and product (9) are recovered. In this scheme, it is possible to utilize a much smaller three phase reactor and provide additional stripping of hydrogen sulfide, if necessary, in a subsequent

step. Such a flow scheme would be beneficial when carrying out the invention in a concurrent flow reactor.

SUMMARY OF THE INVENTION

A method for decreasing sulfur levels in a mercaptan sulfur containing hydrocarbon feedstream comprising the steps of passing said mercaptan sulfur containing hydrocarbon feedstream over a fixed bed catalyst in a three phase, gas, liquid, solid, system in the presence of a stripping gas, for a time and at a temperature and pressure sufficient to decompose at least a portion of said mercaptans to produce olefins, H_2S , as an off gas, and a hydrocarbon product stream having decreased levels of mercaptan sulfur and to disengage said hydrocarbon product stream having decreased amounts of mercaptan sulfur from said H_2S and said stripping gas and wherein when said stripping gas is hydrogen, said fixed catalyst bed comprises (a) a non-reducible metal oxide or (b) a Group VIIIB metal promoted Group VIB catalyst, and wherein when said stripping gas is an inert gas, said fixed bed catalyst comprises a Group VIIIB metal promoted Group VIB catalyst .

As used herein, non-reducible metal oxides are metal oxides that will not reduce to the zero valent metal and water in flowing hydrogen at temperatures below $400^{\circ}C$ and include mixed metal oxides.

As used herein, inert gas, means a gas that is unreactive with unsaturated organics and organosulfur species in the mercaptan sulfur containing feed. The inert gas merely facilitates removal of the H_2S gas produced.

DETAILED DESCRIPTION OF THE INVENTION

An aspect of the invention includes removing mercaptan sulfur from a mercaptan sulfur containing hydrocarbon stream in a three-phase system in the presence of a stripping gas over a mixed metal oxide catalyst.

Thus, a sulfur containing hydrocarbon stream, preferably a previously hydrodesulfurized hydrocarbon stream which still contains an amount of mercaptan sulfur, is passed to a three phase system containing a fixed bed catalyst. The system will be a three-phase system with the catalyst bed located in the hottest zone. The skilled artisan can readily identify the hottest zone for location of the catalyst bed, through, for example, use of thermocouples to read temperature throughout the reactor. The fixed catalyst bed will typically reside at the bottom of the reactor system. Typically, a tray for catalyst is present in such systems and the catalyst will be located in the tray provided.

The mercaptan sulfur containing hydrocarbon stream is reacted over the fixed catalyst bed, in the presence of a stripping gas, to produce H_2S gas and olefins from said sulfur containing hydrocarbon stream. The stripping gas facilitates the disengagement of the hydrocarbon product stream, which will contain the produced olefins, having decreased levels of mercaptan sulfur from the H_2S gas and allows the gases to be removed as off gases from the three phase system.

Any suitable three-phase systems can be employed in the instant invention. For example, a stripper having a fixed catalyst bed in the hottest zone can be employed to accomplish the sulfur removal described herein. Additionally, a fixed bed reactor, such as the one depicted in Figure 3, where the temperature is maintained below the dew point of the hydrocarbon mixture contained within the reactor, so that a substantial portion of the hydrocarbon feed is maintained in the liquid phase may be utilized. Other systems known to the skilled artisan may also be employed. Preferably, in such a system, the

temperature will be maintained at least about 5 °, preferably at least about 10°C below the dew point. Furthermore, the temperature should remain above about 200, preferably above about 250°C. By substantial portion is meant at least about 20%. In operating the invention in this manner, a further stripping step to remove hydrogen sulfide may be employed as shown in Figure 3.

The invention accomplishes the sulfur removal without any significant change in octane of the hydrocarbon stream being acted upon. By significant is meant, no more than about 0.5 number modification in octane number.

The catalyst utilizable for the fixed bed catalyst, when hydrogen is the stripping gas is a non-reducible metal oxide or mixed metal oxide. Non-reducible metal oxides are defined as metal oxides that will not reduce to the zero valent metal and water in flowing hydrogen at temperatures below 400 °C. Non-limiting examples of such oxides include γ -Al₂O₃, SiO₂, SiO₂-Al₂O₃, and MgO and mixtures thereof. If hydrogen is utilized as the stripping gas in the process γ -Al₂O₃ is the preferred catalytic material. Preferably, the catalysts will be sulfided catalysts.

If the stripping gas utilized is an inert gas or hydrogen, the catalyst is a supported group VIIIB metal promoted group VIB catalyst and the inert gas is a non-hydrogenating inert gas. Catalyst examples include, supported and bulk cobalt and nickel promoted molybdenum sulfide catalysts well known in the art, specifically a supported cobalt promoted molybdenum sulfide. Examples of a non-hydrogenating inert gases include nitrogen, helium, argon, methane, natural gas, lighter hydrocarbons in the liquid that are volatilized upon heating (light ends) and mixtures thereof. Light ends are defined as hydrocarbons that have boiling points below the temperature of the reactor.

When hydrogen is utilized as the stripping gas with a hydrogenating catalyst such as CoMoS it is preferred that the amount of stripping hydrogen be minimized. This can be accomplished by minimizing the treat rate of hydrogen. The hydrogen treat rate should be 25 –1000 SCF/B ($4.5 \text{ m}^3/\text{m}^3$ to $180 \text{ m}^3/\text{m}^3$), more preferably 25-500 SCF/B ($4.5 \text{ m}^3/\text{m}^3$ to $90 \text{ m}^3/\text{m}^3$), most preferably 25-250 SCF/B ($4.5 \text{ m}^3/\text{m}^3$ to $45 \text{ m}^3/\text{m}^3$). The hydrogen utilized, can be supplied as part of a gas stream comprising hydrogen, e.g. from a powerformer off gas, thereby leading to a completely integrated refinery process. Notably, the hydrogen stripping gas should contain no more than 1/2 mole percent of H_2S . One skilled in the art will readily recognize that the amount of hydrogen utilized with the Group VIIIB promoted Group VIB catalysts must be controlled to prevent a significant loss of octane. However, by utilizing a hydrogen stream, both mercaptan and thiophenic sulfur can be removed from the hydrocarbon feedstream being acted upon.

Thus, an example of one embodiment of the invention where the mercaptan sulfur containing feedstream is a hydrosulfurized (SCANfined) feedstream is depicted in the figure 2. Olefinic naphtha (Stream 1), such as catalytic cracked and steam cracked naphtha are mixed with hydrogen (Stream 2) and reacted in a selective naphtha hydrofining reactor (Reactor 12). The organic sulfur compounds in the olefinic naphtha feed are predominantly thiophenes. The vapor product of the selective naphtha (Stream 3) contains significantly lower levels of thiophenic sulfur and hydrogen sulfide but still contains significant quantities of olefins and mercaptan sulfur. The mercaptans in Stream 3 are produced through reaction of product hydrogen sulfide with feed olefins. Stream 3 is then cooled in Heat Exchanger 13 such that the C5+ fraction is liquefied in Separation Drum 14. The overhead sour gas stream (Stream 5) which contains unreacted hydrogen and the majority of the product hydrogen sulfide is sent to Scrubber vessel 15 where hydrogen sulfide is removed to produce a sweet hydrogen stream (Stream 6). Stream 6 is compressed in Compressor 16 to the

operating pressure of the Reactor 12 where it is utilized for the hydroprocessing reaction. The liquid product (Stream 7) from Separation Drum 14, contains lower levels of organo sulfur, both as thiophenes and mercaptans, in addition to olefins, paraffins, aromatics, and dissolved hydrogen sulfide. Though, in a typical process, this stream would be depressurized through a pressure relief (number 17 in Figure 1) and sent to a stripper (18), the depressurization step is omitted in the embodiment of the process shown in Figure 2 process. In stripper 18 stream 7 is contacted with an inert gas over a fixed bed catalyst to produce hydrogen sulfide, which exits with the inert gas in stream 10. The stripper reactor may be filled with catalyst coated packings, but it is preferred that the catalyst be loaded onto bubble trays in order to maximize residence time in the reaction zone of the stripper.

Another embodiment of this invention would be the use of a three-phase fixed bed reactor shown as Reactor 19 in Figure 3. In this process the liquefied desulfurized naphtha (Stream 7) is reacted with a hydrogen or non-hydrogenating stripping gas in a fixed bed reactor containing catalyst. The temperature of this reactor is maintained at a temperature below the dew point of the feed mixture. The product of this reactor (Stream 20) is depressurized (Pressure let down 17) followed by removal of dissolved hydrogen sulfide here in a stripper 18. Alternatively, a flash drum could be used in place of the stripper, for example.

The three phase reactor system of the invention (in the example above, a stripper) is operated at pressures of at least about 115 psi (791 kPa), more preferably greater than 150 psi (1034 kPa), and most preferably greater than 200 psi such that the temperature of bottom of the vessel where the catalyst bed will be located is established by the boiling point of the heaviest components in the feed at the pressure of the vessel. The higher the pressure the higher the temperature of the catalyst zone. It is preferred that the catalyst zone temperature

be above 200 °C, more preferably above 225 °C and most preferably above 250 °C. Preferably, the temperature will not exceed 400 °C. The amount of stripping gas added should not exceed the amount that would increase the dew point of the reactor to a temperature below that of the desired operating temperature. The gas flow rates would typically be 25 to 750 SCF/B (4.5 to 139 m³/m³), more preferably 25 to 500 SCF/B (4.5 to 90 m³/m³). The conditions selected favor mercaptan destruction kinetics and thermodynamics.

In a preferred embodiment, the three-phase reactor system is operated in a concurrent or counter current fashion with the countercurrent fashion being preferred. In counter current mode the liquid and gas move in opposite directions of each other. Typically liquid is injected in the top or middle of the vessel and flows downward exiting the bottom of the vessel. Gas is injected in the bottom of the reactor and moves upward through the liquid phase, thereby stripping dissolved gaseous components, exiting through the top of the vessel.

Because the selective removal or conversion of mercaptans from a previously hydrodesulfurized hydrocarbon stream is readily accomplished by the instant invention, it is possible to operate the HDS unit to achieve a higher total sulfur level, thereby preserving feed olefins and octane and then perform the method of the invention to remove the mercaptans affording an integrated process for producing a high quality product. Hence, less severe HDS conditions can be employed when an HDS step is coupled with the process herein described since the mercaptans from the HDS process can be readily decreased or removed in the process.

For example, an intermediate cat naphtha can be hydroprocessed to 60 wppm total sulfur where approximately 45 wppm sulfur is mercaptan sulfur. This first product would not meet the future 30 wppm sulfur specification. This

product would then be treated with the method of sulfur removal described herein as the mercaptan sulfur containing feedstream in a three phase reactor with a fixed bed catalyst where the sulfur level would be reduced to approximately 20 wppm total sulfur, meeting environmental specifications. By not hydroprocessing the intermediate cat naphtha directly to 20 wppm sulfur, olefin saturation will be less than is obtained from hydroprocessing to 20 wppm directly. Thus, considerable octane is preserved affording an economical and regulatory acceptable product.

If it is desired to hydrodesulfurize the sulfur containing feedstream prior to passing it to the three-phase reactor with fixed bed catalyst described herein, any hydrodesulfurization process known in the art can be utilized.

Preferably, the feedstream to the three phase reactor will have less than 30 ppm of non-mercaptan sulfur, more preferably the feedstream will have less than 30 ppm non-mercaptan sulfur and greater than 30 ppm of mercaptan sulfur. Any hydrodesulfurization step capable of producing such feedstreams can be conducted prior to the three phase reactor process herein described and the resultant product sent to the three-phase reactor.

If it is desired to hydrodesulfurize the mercaptan sulfur containing hydrocarbon feedstream to produce increased quantities of mercaptans and retain olefins prior to the treatment herein, any technique known in the art can be utilized. For example, the hydrotreated hydrocarbon stream can be hydrodesulfurized to produce a sulfur containing hydrocarbon stream which contains non-mercaptan sulfur at a level below the mogas specification as well as significant amounts of mercaptan sulfur. Generally, such processing conditions will fall within the following ranges: 475-600 °F (246-316 °C), 150-500 psig (1136-3548 kPa) total pressure, 100-300 psig (791-2170 kPa) hydrogen partial pressure, 1000-2500 SCF/B hydrogen treat gas, and 1-10 LHSV.

The preferred hydroprocessing step to be utilized if prior HDS is desired, is SCANfining. However, other selective cat naphtha hydrodesulfurization processes such as those taught by Mitsubishi (see US patents 5,853,570 and 5,906,730) can likewise be utilized herein. SCANFINING is described in National Petroleum Refiners Association paper # AM-99-31 titled "Selective Cat Naphtha Hydrofining with Minimal Octane Loss" and US patents 5,985,136 and 6,013,598 herein incorporated by reference. Selective cat naphtha HDS is also described in US patents 4,243,519 and 4,131,537.

Typical SCANfining conditions include one and two stage processes for hydrodesulfurizing a naphtha feedstock comprising reacting said feedstock in a first reaction stage under hydrodesulfurization conditions in contact with a catalyst comprised of about 1 to 10 wt. % MoO_3 ; and about 0.1 to 5 wt. % CoO ; and a Co/Mo atomic ratio of about 0.1 to 1.0; and a median pore diameter of about 60 [Angstrom] to 200 [Angstrom]; and a MoO_3 surface concentration in $\text{g MoO}_3/\text{m}^2$ of about 0.5×10^{-4} to 3×10^{-4} ; and an average particle size diameter of less than about 2.0 mm; and, optionally, passing the reaction product of the first stage to a second stage, also operated under hydrodesulfurization conditions, and in contact with a catalyst comprised of at least one Group VIII metal selected from the group consisting of Co and Ni, and at least one Group VI metal selected from the group consisting of Mo and W, more preferably Mo, on an inorganic oxide support material such as alumina.

The SCANFINING reactor can be run at sufficient conditions such that the difference between the total organic sulfur (determined by x-ray adsorption) and the mercaptan sulfur (determined by potentiometric test ASTM3227) is at or below the desired (target) specification (typically 30 ppm for non-mercaptan sulfur). This stream is then sent to the three-phase system described herein for mercaptan removal.

The three phase system method described herein is particularly capable of removing $\geq C_5+$ mercaptan sulfur.

The product from the instant process is suitable for blending to make motor gasoline that meets sulfur specifications of ≤ 30 ppm range.

The following examples, which are meant to be illustrative and not limiting, illustrate the potential benefit of the invention, by showing specific cases in which a selective hydrofining process has been operated to produce varying levels of total and mercaptan sulfur. By reference to these cases, it should be apparent that coupling such selective hydrotreating with a subsequent mercaptan removal technology will result in improved ability to produce low sulfur products with reduced losses of olefins and octane.

The following example is illustrative and not meant to be limiting.

Example 1

A flow through catalytic test was conducted to test catalytic materials and stripping gases for a catalytic stripping reactor. A fixed bed reactor was loaded with either 5 cc of a commercial γ - Al_2O_3 or a commercial cobalt promoted molybdenum sulfide hydrotreating catalyst (CoMoS). The cobalt promoted molybdenum catalyst was pre-sulfided with hydrogen sulfide. A previously hydrotreated intermediate cat naphtha feed was reacted over these two catalysts at 260 °C, 235 psia pressure (1620 kPa), 2.0 LHSV, and with 675 SCF/B ($122 \text{ m}^3/\text{m}^3$) of either hydrogen or nitrogen. These conditions were chosen to mimic those in the bottom of a high pressure-stripping reactor. The hydrotreated intermediate cat naphtha had a total sulfur of 60 wppm and mercaptan sulfur content of 43 wppm, and a bromine number of 20. As can be seen in the table below, alumina with hydrogen as the stripping gas results in 56% conversion of

the mercaptan sulfur with no saturation of the olefins. When nitrogen is used as the stripping gas alumina rapidly deactivates and almost no conversion is observed. The CoMoS catalyst with hydrogen removes almost all the sulfur including some of the non-mercaptan sulfur, but undesirable saturation of the olefins is observed. When nitrogen is used as the stripping gas CoMoS removes approximately 95% of the mercaptan sulfur and no undesired olefin saturation is observed. CoMoS showed no apparent deactivation with nitrogen as the stripping gas.

Table

Catalyst	γ -Al ₂ O ₃	γ -Al ₂ O ₃	CoMoS	CoMoS
Stripping Gas	H ₂	N ₂	H ₂	N ₂
wppm Sulfur (XRF)	36	53	5	20
Bromine Number	18.5	18.8	12.8	18.7

WHAT IS CLAIMED IS:

1. A method for decreasing sulfur levels in a mercaptan sulfur containing hydrocarbon feedstream comprising the steps of passing said mercaptan sulfur containing hydrocarbon feedstream over a fixed bed catalyst in a three phase, gas, liquid, solid, system in the presence of a stripping gas, for a time and at a temperature and pressure sufficient to decompose at least a portion of said mercaptans to produce olefins H_2S , as an off gas, and a hydrocarbon product stream having decreased levels of mercaptan sulfur and to disengage said hydrocarbon product stream having decreased amounts of mercaptan sulfur from said H_2S and said stripping gas and wherein said stripping gas is hydrogen, said fixed catalyst bed comprises (a) a non-reducible metal oxide or (b) a Group VIIIB metal promoted Group VIB catalyst, and wherein when said stripping gas is an inert gas or hydrogen, said fixed bed catalyst comprises a Group VIIIB metal promoted Group VIB catalyst.
2. The method of claim 1 wherein said inert gas is selected from helium, nitrogen, argon, methane, natural gas, light ends and mixtures thereof.
3. The method of claim 1 wherein said non-reducible metal oxide catalyst is selected from alumina, silica-alumina, magnesium oxide, and mixtures thereof and said Group VIIIB promoted Group VIB catalyst is selected from the group consisting of cobalt, and nickel promoted molybdenum catalysts.
4. The method of claim 2 wherein when said stripping gas is gas comprising hydrogen and said catalyst is a Group VIIIB promoted Group VIB catalyst, said stripping gas comprises no more than 1/2 mole % hydrogen sulfide and no more than 50 mole % hydrogen.

5. The method of claim 1 wherein said mercaptan sulfur containing hydrocarbon feedstream is a hydrodesulfurized feedstream.

6. The method of claim 1 wherein said method includes a hydrodesulfurization step to produce said mercaptan sulfur containing hydrocarbon feedstream.

7. The method of claim 6 wherein said hydrodesulfurization step is SCANfining.

8. The method of claim 1 wherein said mercaptan sulfur containing hydrocarbon feedstream is a C_5^+ mercaptan containing feedstream.

9. The method of claim 1 wherein said three phase system is a countercurrent system.

10. The method of claim 1 wherein said three phase system is a concurrent system.

11. The method of claim 3 wherein said catalysts are sulfided catalysts.

12. The method of claim 1 wherein said mercaptan sulfur containing hydrocarbon feedstream contains less than 30 ppm of non-mercaptan sulfur.

13. The method of claim 1 wherein said mercaptan sulfur containing hydrocarbon feedstream contains less than 30 ppm of non-mercaptan sulfur and greater than 30 ppm of mercaptan sulfur.

14. The method of claim 13 wherein said mercaptan sulfur containing feedstream is produced from a hydrodesulfurization process.

4

[illegible]

Figure 1

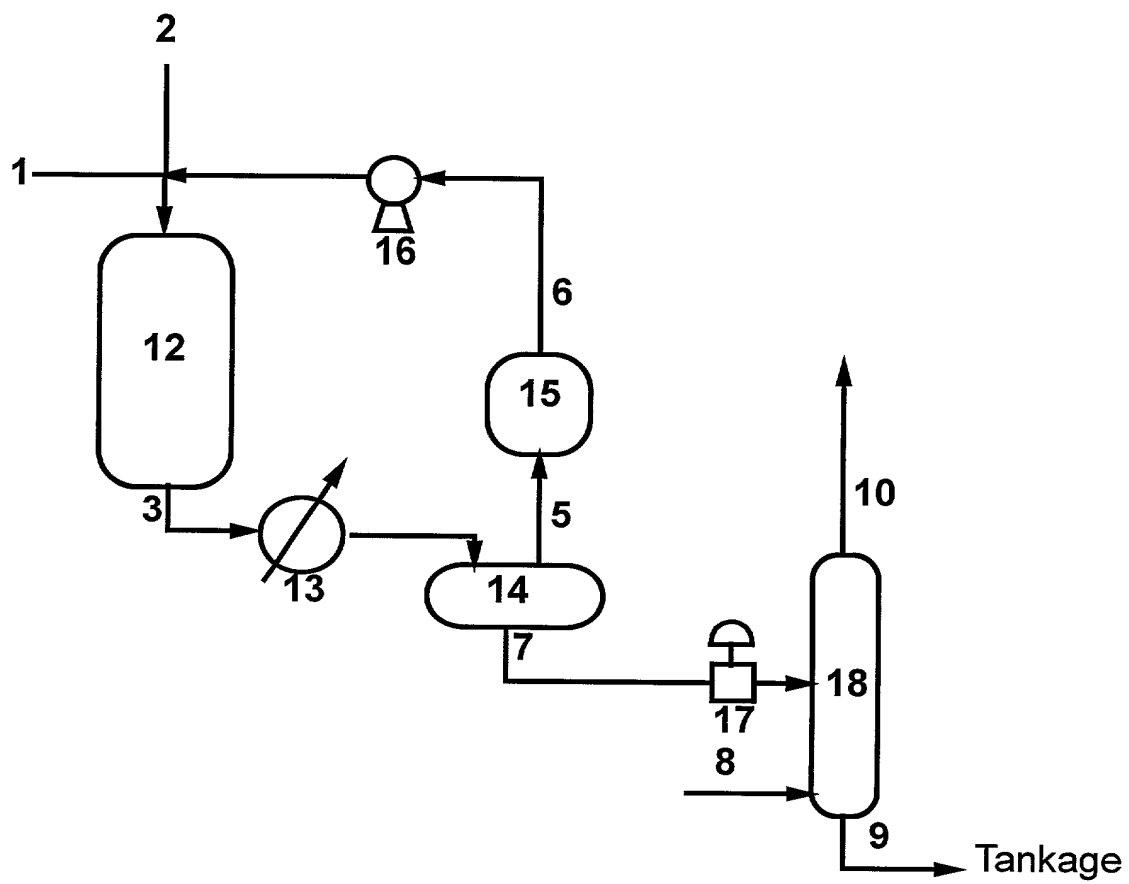


Figure 2

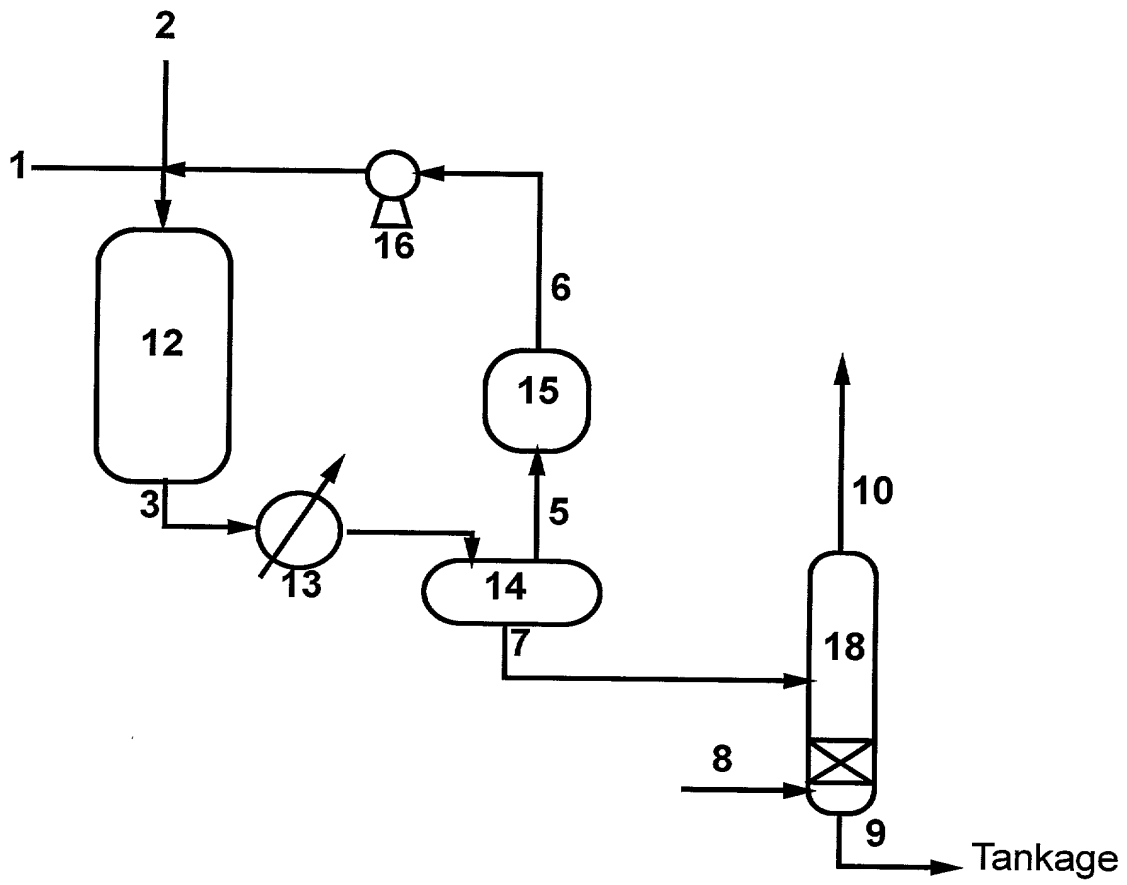
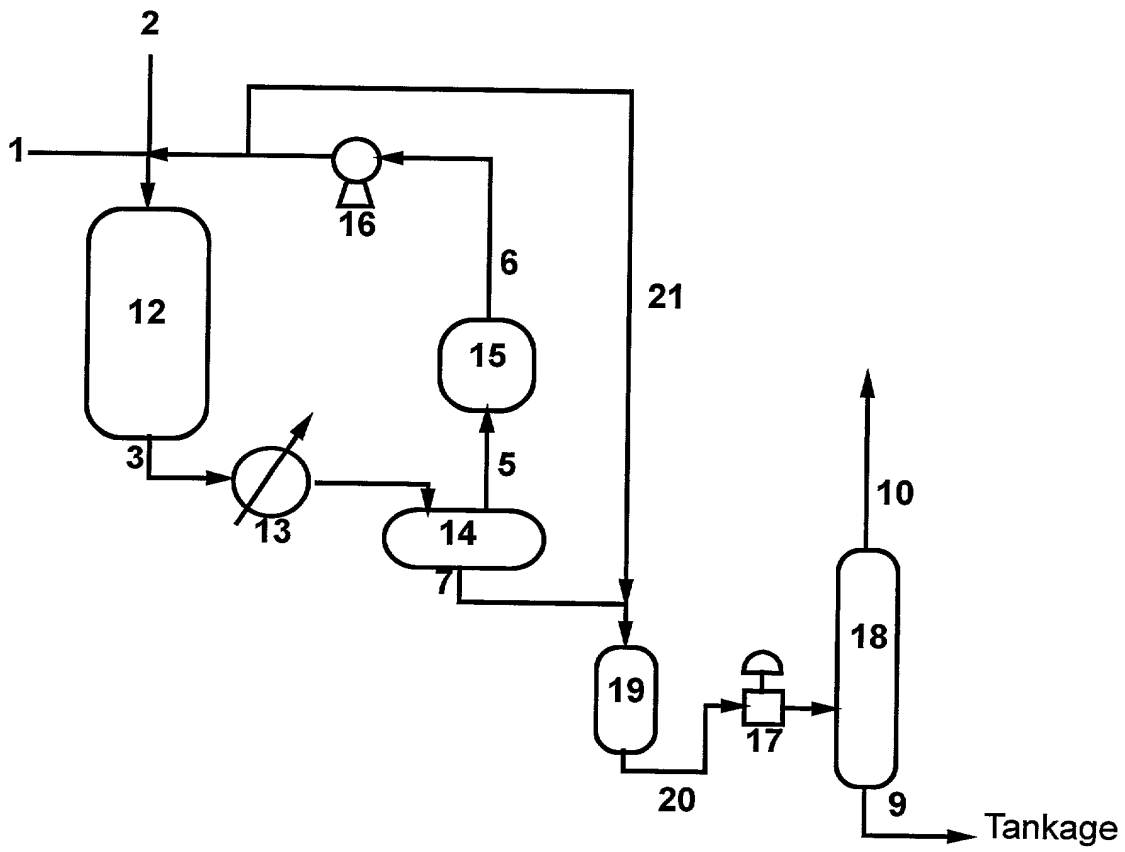


Figure 3



DECLARATION FOR PATENT APPLICATION

Case Docket No. ECB-0004

As below named inventor(s)

BRUCE R. COOK
WILLIAM E. WINTER, JR.

ROBERT C. WELCH
RICHARD A. DEMMIN

RICHARD H. ERNST
CHARLES A. EUKER, JR.

We/I hereby declare that residence, post office address and citizenship are as stated on page 2.

We/I believe we/I are the original, first and joint inventor(s) of the subject matter which is claimed and for which a patent is sought on the invention entitled CATALYTIC STRIPPING FOR MERCAPTAN REMOVAL, the specification of which

(check one) ☒ is attached hereto.

☐ was filed on _____ as Application Serial No. _____
and was amended on _____ (if applicable).

We/I hereby state that we/I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

We/I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

We/I hereby claim foreign priority benefits under Title 35, United States Code, §119 of any foreign application(s) for patent or inventor's certificate(s) listed below and have also identified below any foreign application(s) for patent or inventor's certificate(s) having a filing date before that of the application on which priority is claimed:

Prior Foreign Application(s)

			Priority Claimed	
_____ (Number)	_____ (Country)	_____ (Day/Month/Year Filed)	<input type="checkbox"/> Yes	<input type="checkbox"/> No

I hereby claim the benefit under 35 U.S.C. § 119(e) of any United States provisional application(s) listed below.

(Application Number)	(Filing Date)
_____	_____

We/I hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, we/I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

_____ (Application Serial No.)	_____ (Filing Date)	_____ (Status) (patented, pending, abandoned)
_____ (Application Serial No.)	_____ (Filing Date)	_____ (Status) (patented, pending, abandoned)

POWER OF ATTORNEY: As named inventor(s), we/I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith.

NAMES	REGISTRATION NUMBERS
ESTELLE C. BAKUN	35,054
PAUL E. PURWIN	29,203

SEND CORRESPONDENCE TO:

EXXONMOBIL RESEARCH AND ENGINEERING COMPANY
(formerly Exxon Research and Engineering Company)
P. O. Box 900 / Clinton Township
Annandale, New Jersey 08801-0900

DIRECT TELEPHONE CALLS TO: (Name and Telephone Number)

ESTELLE C. BAKUN
(908) 730-3635

RECEIVED

SEP 27 2000

Case Docket No. ECB-0004

EMRE LAW DEPT

We/I hereby declare that all statements made herein of our/my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

FULL NAME OF INVENTOR	LAST NAME COOK	FIRST NAME BRUCE	MIDDLE NAME RANDALL
RESIDENCE & CITIZENSHIP	CITY STEWARTSVILLE	STATE OR FOREIGN COUNTRY NEW JERSEY	COUNTRY OF CITIZENSHIP U.S.A.
POST OFFICE ADDRESS	POST OFFICE ADDRESS 1101 MONROE DRIVE	CITY STEWARTSVILLE	STATE OR COUNTRY NEW JERSEY
			ZIP CODE 08886

Inventor's Signature



Date

8/7/00

FULL NAME OF INVENTOR	LAST NAME WELCH	FIRST NAME ROBERT	MIDDLE NAME CHARLES
RESIDENCE & CITIZENSHIP	CITY BATON ROUGE	STATE OR FOREIGN COUNTRY LOUISIANA	COUNTRY OF CITIZENSHIP U.S.A.
POST OFFICE ADDRESS	POST OFFICE ADDRESS 542 WOODRUFF DRIVE	CITY BATON ROUGE	STATE OR COUNTRY LOUISIANA
			ZIP CODE 70808-5122

Inventor's Signature




Date

8/29/00

FULL NAME OF INVENTOR	LAST NAME ERNST	FIRST NAME RICHARD	MIDDLE NAME HARRY
RESIDENCE & CITIZENSHIP	CITY GLEN GARDNER	STATE OR FOREIGN COUNTRY NEW JERSEY	COUNTRY OF CITIZENSHIP U.S.A.
POST OFFICE ADDRESS	POST OFFICE ADDRESS 125 HOCKENBURY DRIVE	CITY GLEN GARDNER	STATE OR COUNTRY NEW JERSEY
			ZIP CODE 08826-3520

Inventor's Signature

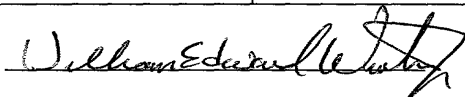


Date

8/2/00

FULL NAME OF INVENTOR	LAST NAME WINTER, JR.	FIRST NAME WILLIAM	MIDDLE NAME EDWARD
RESIDENCE & CITIZENSHIP	CITY BATON ROUGE	STATE OR FOREIGN COUNTRY LOUISIANA	COUNTRY OF CITIZENSHIP U.S.A.
POST OFFICE ADDRESS	POST OFFICE ADDRESS 1026 MARC ANTONY DRIVE	CITY BATON ROUGE	STATE OR COUNTRY LOUISIANA
			ZIP CODE 70816

Inventor's Signature

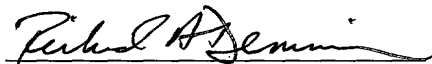


Date

8/29/2000

FULL NAME OF INVENTOR	LAST NAME DEMMIN	FIRST NAME RICHARD	MIDDLE NAME ALAN
RESIDENCE & CITIZENSHIP	CITY HIGHLAND PARK	STATE OR FOREIGN COUNTRY NEW JERSEY	COUNTRY OF CITIZENSHIP U.S.A.
POST OFFICE ADDRESS	POST OFFICE ADDRESS 424 DENNISON STREET	CITY HIGHLAND PARK	STATE OR COUNTRY NEW JERSEY
			ZIP CODE 08904

Inventor's Signature

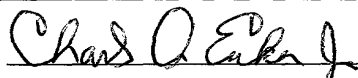


Date

9/7/2000

FULL NAME OF INVENTOR	LAST NAME EUKER, JR.	FIRST NAME CHARLES	MIDDLE NAME ALFRED
RESIDENCE & CITIZENSHIP	CITY BATON ROUGE	STATE OR FOREIGN COUNTRY LOUISIANA	COUNTRY OF CITIZENSHIP U.S.A.
POST OFFICE ADDRESS	POST OFFICE ADDRESS 5130 BELLE FONTAINE COURT	CITY BATON ROUGE	STATE OR COUNTRY LOUISIANA
			ZIP CODE 70820

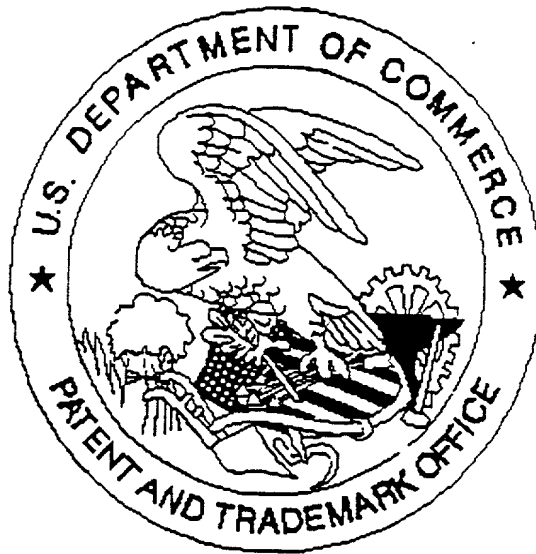
Inventor's Signature



Date

9/21/00

United States Patent & Trademark Office
Office of Initial Patent Examination -- Scanning Division



Application deficiencies were found during scanning:

☐ Page(s) _____ of _____ were not present
for scanning. (Document title)

☐ Page(s) _____ of _____ were not present
for scanning. (Document title)

☒ Scanned copy is best available.

Declaration Page 1-2 are torn
by corner.